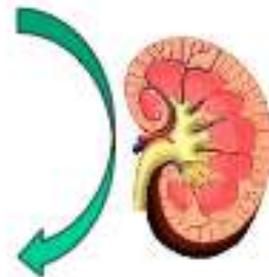
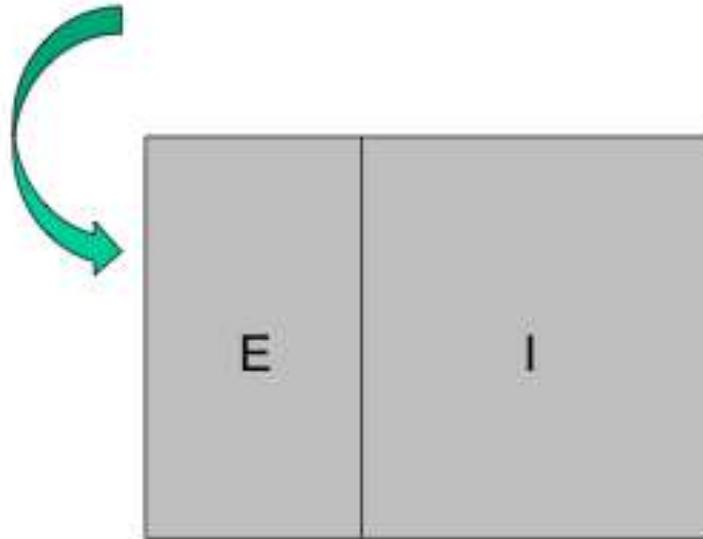


# Misadventures in salt & water, as well as in acid-base balance

Entertaining you is  
Friedrich C. Luft, Berlin

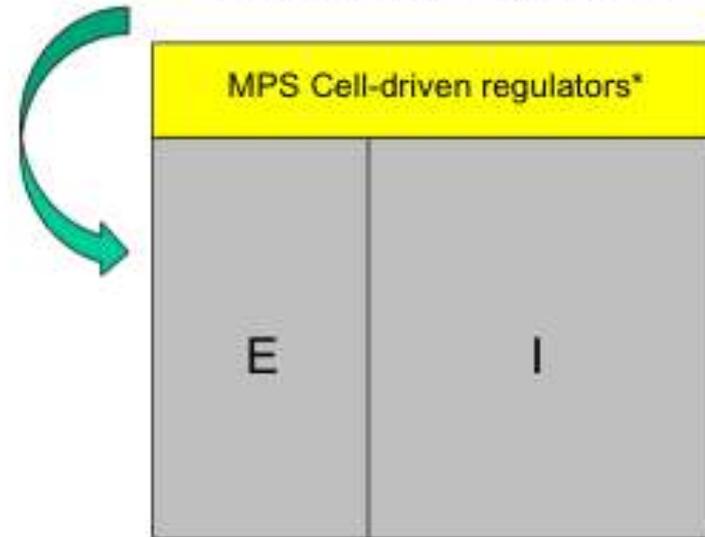
*Charité*

We teach this:

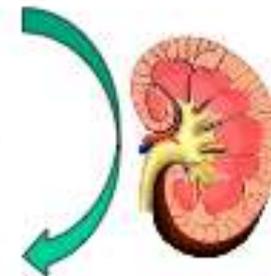


We found this:

Third compartment NaCl in skin, muscle, and elsewhere

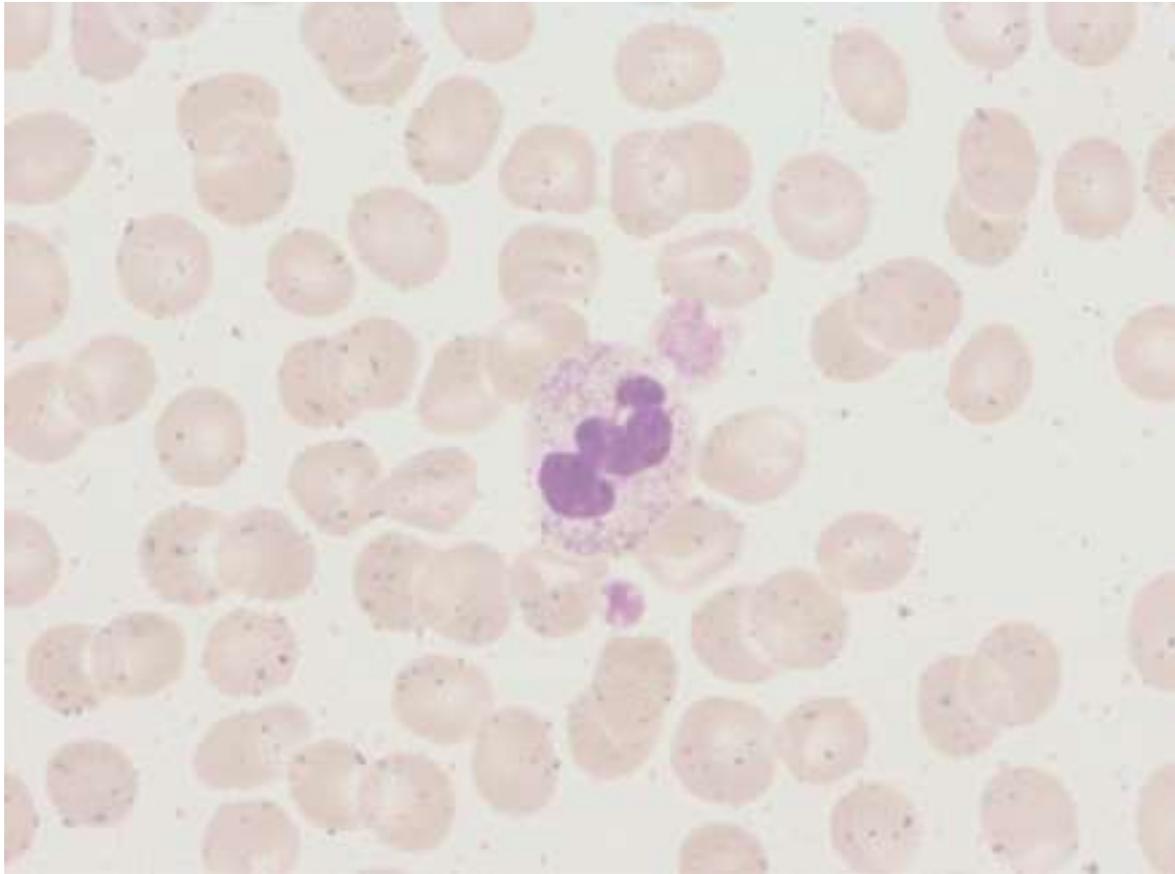


\*VEGF-C  
VEGF-C sR  
(sFlt4)



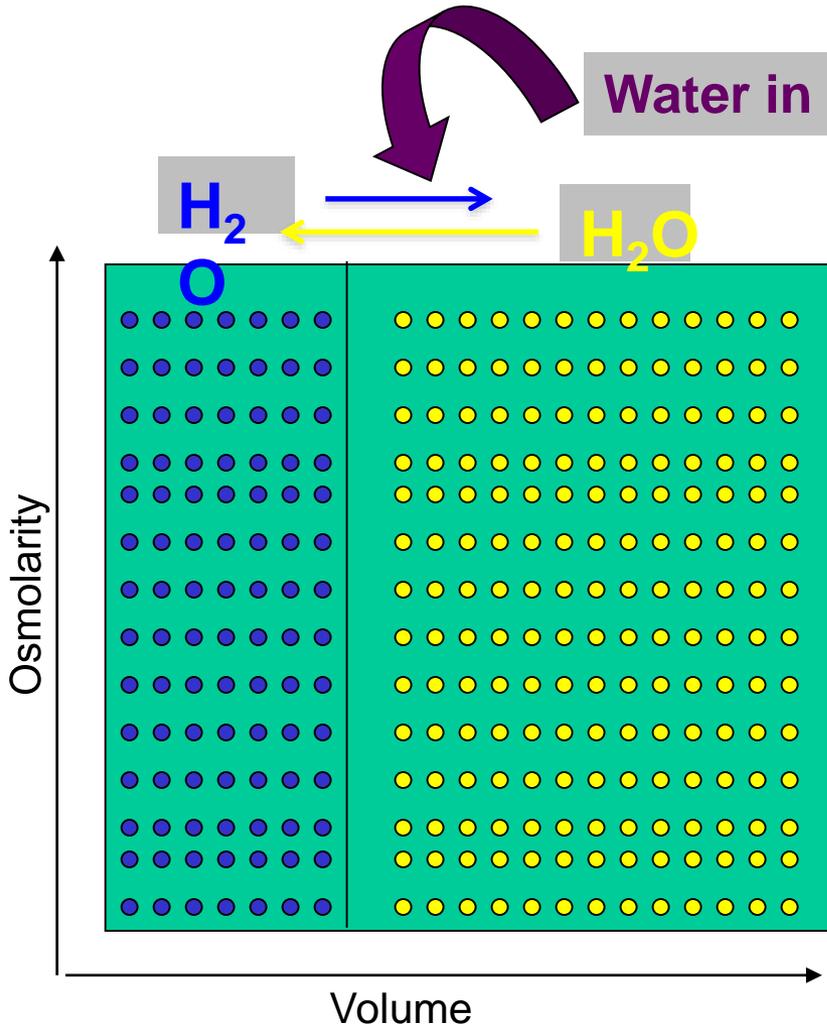
# Don't just “do something” – stand there

- 68 year-old woman presents disoriented at 18:00; had undergone tooth extraction that morning and, aside from a life-long mild bleeding tendency, had been quite normal
- BP 130/85, pulse regular, respirations 18/min no localizing findings, no edema
- Na 118, K 3.6, glucose 8, urea 4 (all mmol/L)
- What now?



An oil-immersion field showing a normal neutrophil flanked by two giant platelets (Bernard-Soulier syndrome). She had been given **desmopressin**. In addition, it had been hot so she was advised to “**drink lots of water**”

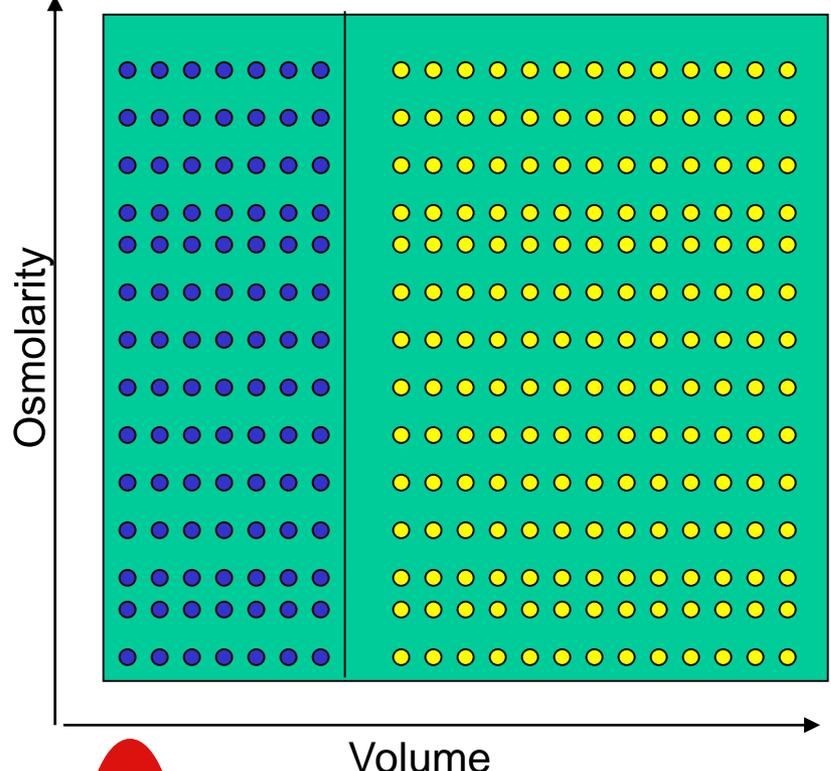
# Serum-Na depends on TBW, Na and K



- Na
- K

$$\text{Serum Na} \approx \frac{\text{Na}_{\text{exch}} + \text{K}_{\text{exch}}}{\text{Total-body H}_2\text{O}}$$

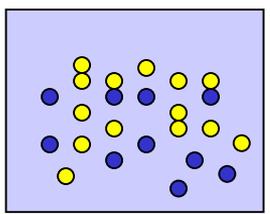
Edelman formula



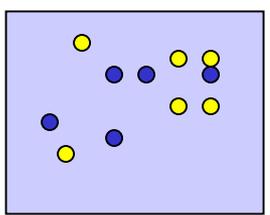
● Na  
● K

$$\text{Serum Na} \approx \frac{\text{Na}_{\text{exch}} + \text{K}_{\text{exch}}}{\text{total-body H}_2\text{O}}$$

$$\text{Clearance H}_2\text{O (e)} = V \left[ 1 - \frac{U_{\text{Na}+\text{K}}}{S_{\text{Na}}} \right]$$



Lots of spheres  
= little H<sub>2</sub>O  
Cl<sub>H<sub>2</sub>O</sub>(e) neg



Few spheres  
= much H<sub>2</sub>O  
Cl<sub>H<sub>2</sub>O</sub>(e) pos



When  $U_{\text{Na}+\text{K}} > S_{\text{Na}}$  the Cl<sub>H<sub>2</sub>O</sub>(e) neg and serum Na must fall

When  $U_{\text{Na}+\text{K}} < S_{\text{Na}}$ , the Cl<sub>H<sub>2</sub>O</sub>(e) pos and serum Na must rise

Actually, serum Na increased a little faster than we wanted so we infused some free water

**Table 1.**

Serum and urine values (mmol/L) over a timeframe of 38 h

<b>Time</b>	<b>18.00</b>	<b>06:00</b>	<b>21:00</b>	<b>08:00</b>
			Glucose 5% infusion	
Serum				
Sodium	118	125	138	138
Potassium	3.6	3.0	5.4	4.4
Urine				
Sodium	16	-	24	43
Potassium	7	-	9	16

Had we given 3% saline, serum Na would have increased even faster

## Iatrogenic SIADH

# Paradoxal hyponatremia with isotonic electrolyte infusions

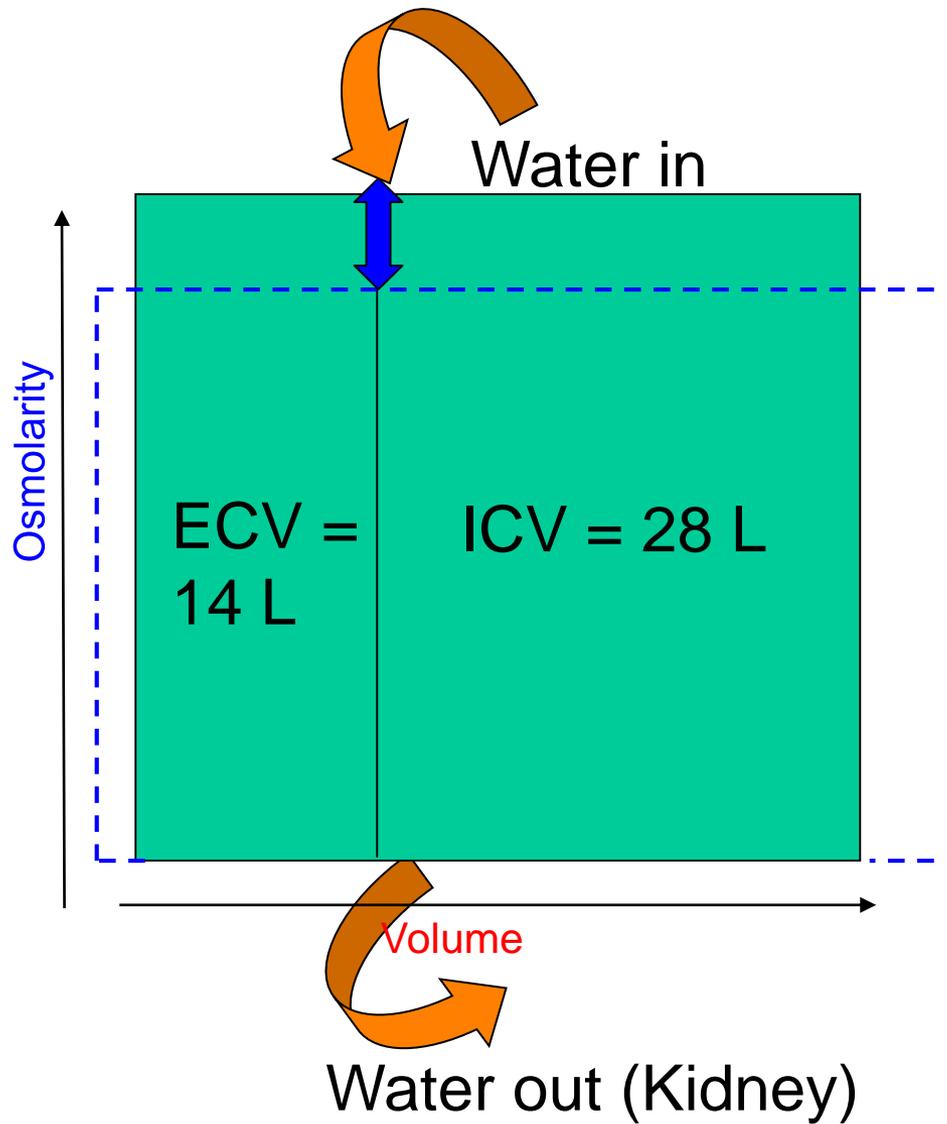
- 65 year-old woman has meniscus surgery. At that time her Na was 141 mmol/L. She is hypertensive and receives HCTZ and ACEI, as well as NSAID + opiates for pain
- Lots of nausea and vomiting post-op
- Two days later Na 120, K 3.2 in emergency
- She receives Ionosteril (Na 137) + KCl 20 mmol/l
- Two days later Na 116, K 2.8. Same therapy given again

# How is it that the serum Na got worse?

- Shortly thereafter, admitted with Na 98, K 2.3, glucose 6, urea 3 (all mmol/L) and grand-mal seizure
- Exam BP 200/110 mm Hg, patient is post-ictal (comatose)
- She has no edema
- What is her plasma osmolarity?

$$(98+2.3)\times 2 + 6 + 3 = ?$$





**At bwt 70 kg →  
12 L Excess!!**

$$\text{Water excess} = 0.6 \times \text{Bwt (kg)} \times (1 - \frac{\text{Na measured}}{140})$$

# Appropriate reasoning with hyponatremic patients who have high ADH ( $U_{osmo} > 300$ )

What caused ADH release?

Nonosmotic stimuli with volume disturbance

- Low ECV
- High ECV (Edema)  
(Heart-, Liver-, Kidney, Adrenal  
You will find them here)

Nonosmotic stimuli with euvolemia (SIADH)

Euvolemia, with  
diseases or drugs  
that cause ADH  
release

Gotcha!



# Who is teaching fluid and electrolytes?

A 22 year-old woman comes to emergency because of weakness, vomiting and abdominal discomfort. She has felt poorly several weeks but recently worse. She has Hashimoto's thyroiditis and receives replacement

Hb 14 g/dL, Na 122, K 4.8, Cl 92, glucose 2.5, urea 5 (all mmol/L). Creatinine 0.7 mg/dL. The pH was 7.17, PaO<sub>2</sub> 95, PaCO<sub>2</sub> 28 (mm Hg), HCO<sub>3</sub> 11

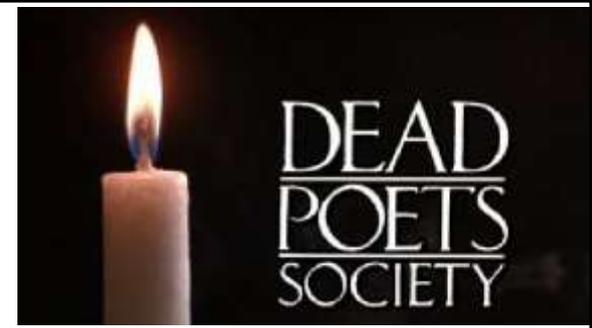
Ultrasound indicates that appendicitis "cannot be ruled out"



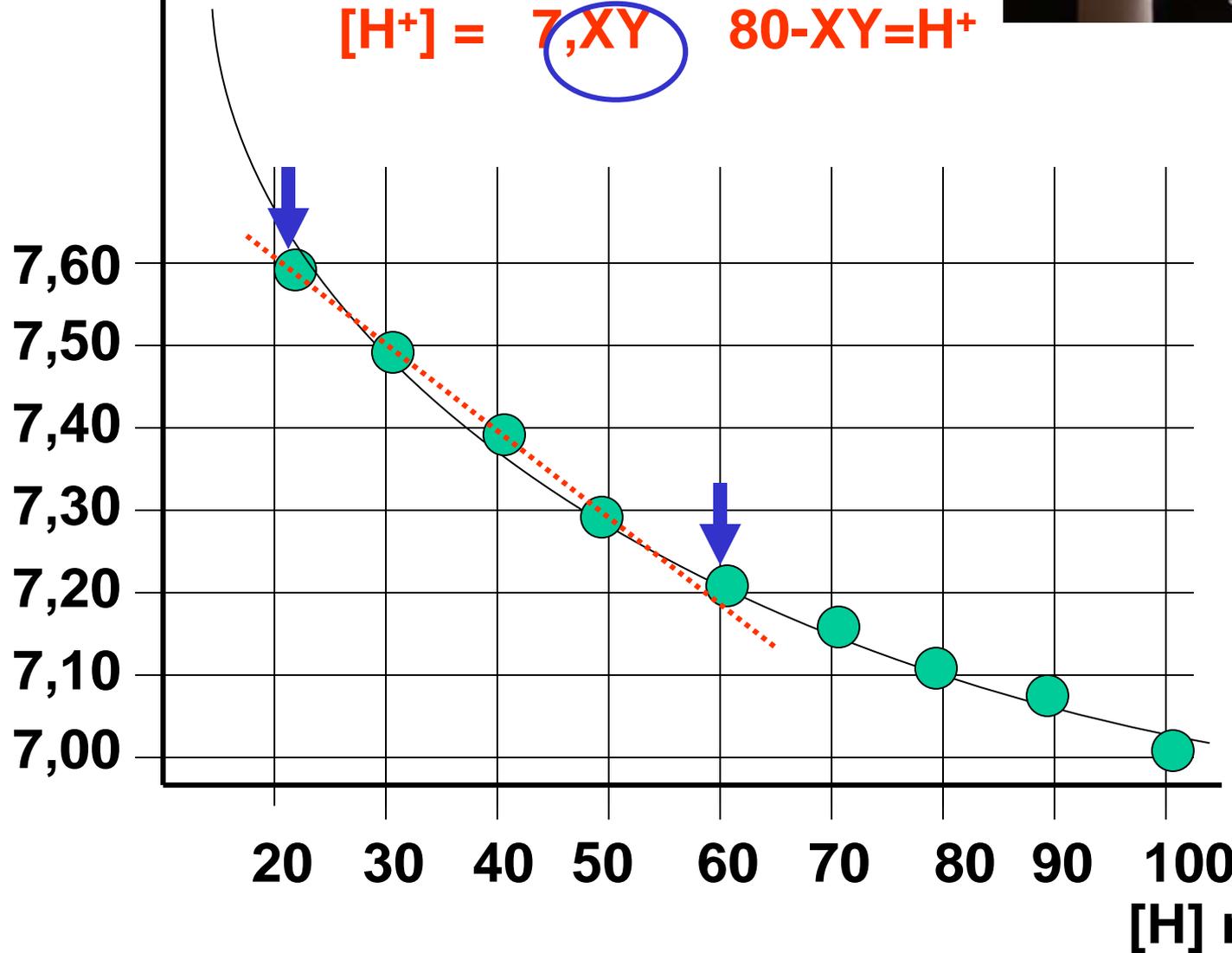
pH  
Log 1/[H]

# „puissance“ Hydrogen

Sorensen & Hasselbalch

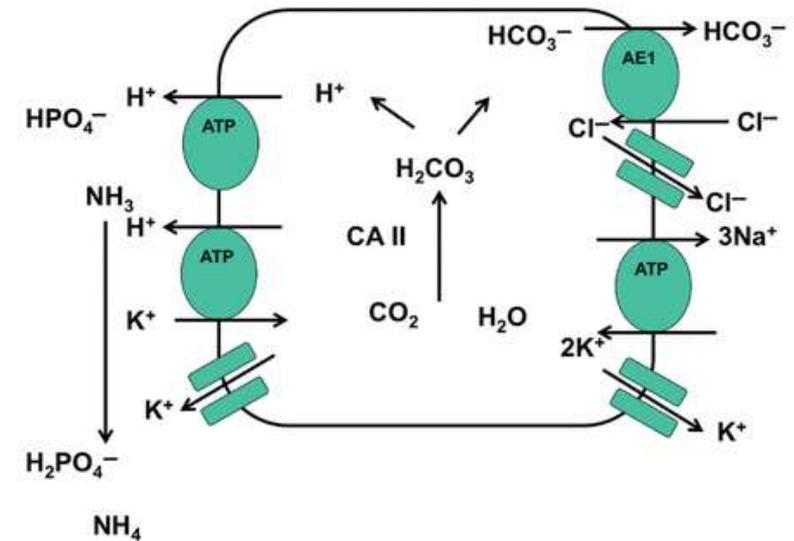


$$[H^+] = 7,XY \quad 80-XY=H^+$$



Would have  
fired them all  
on the spot!

$$[H] = K \frac{H_2CO_3}{HCO_3^-} \quad (K = \text{Dissoziationskonstante})$$

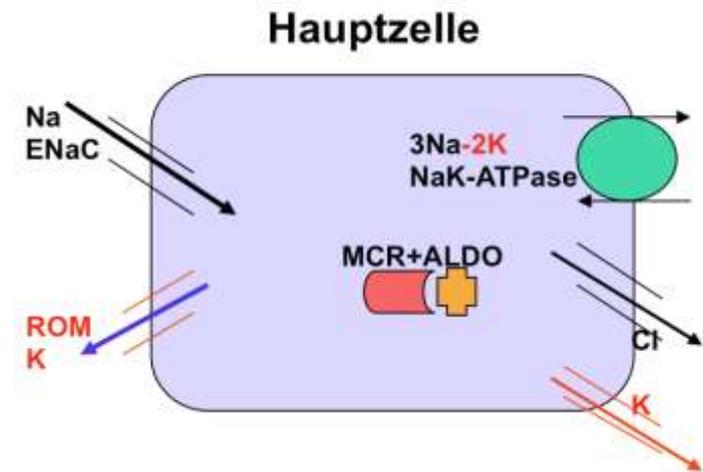


$$H = (24 \times PaCO_2) / HCO_3^-$$

$$H = (24 \times 28) / 11 = 61 \text{ (nmol/L)}$$

$$AG = 122 - (92 + 11) = 19$$

etc



Next day, we are called because she is no better.  
BP 105/70 mm Hg. She becomes unconscious when held upright. Her blood labs are unchanged.

Urine Na 135, K 29, Cl 69 (mmol/L), Uosm 644

Stat cortisol 3.1  $\mu\text{g/dL}$ , after ACTH 3.2  $\mu\text{g/dL}$

Aldo (took a few days) "low"

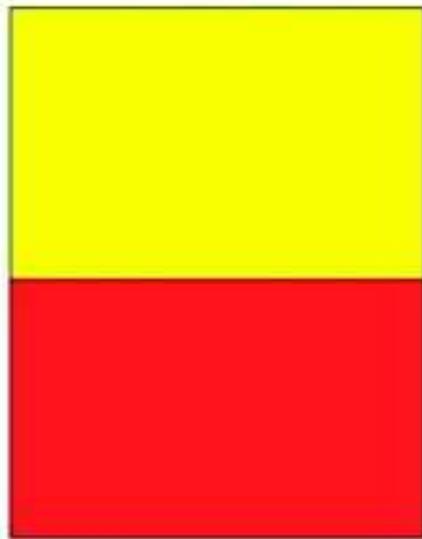
ACTH 1250 (normal  $<40$ )

Only half of Addison's disease patients have hyperkalemia

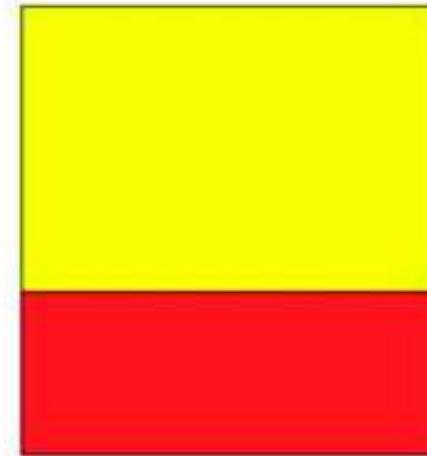
After hydrocortisone and several liters ionosteril her Hemoglobin dropped to 10 g/dL without any blood loss

Hct 45 vol%

Hct 30 vol%



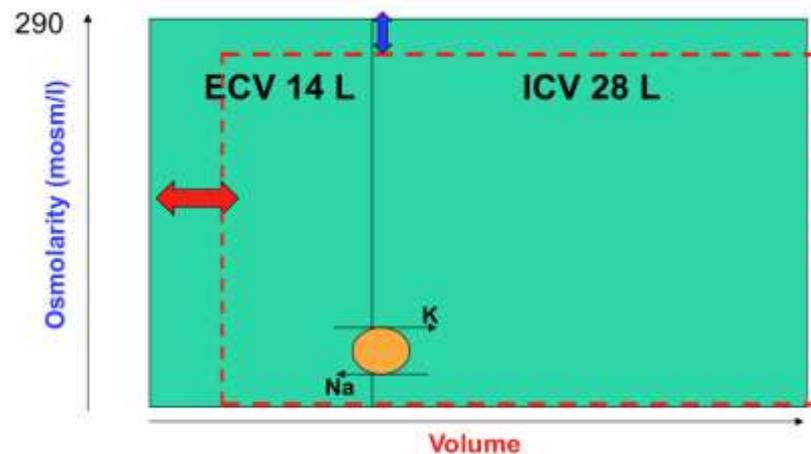
Hct 45 vol%



Should have been

Likely was

Then became



**Addison's disease (polyendocrine syndrome)**

# Andy is a young Canadian visiting Brandenburg

His ancestors had emigrated on the “Hopewell” in 1761  
Andy runs his Kawasaki into a tree and lands with our neurosurgeons. They drain an epidural hematoma.  
Post op they call us because Andy “pees a lot”.

Na 166, Cl 132, K 4, HCO<sub>3</sub> 22 (mmol/L)

Crea 230 μmol/L, urea 10, glucose 5 (mmol/L)

Urine output 8 L/24 h

IV input around 7 L/24 h

Diuresis?

Water or solute?

Can Andy concentrate his urine?

What does he require?

## Solute diuresis

Glucose

Urea

Mannitol

Xylite

NaCl or other

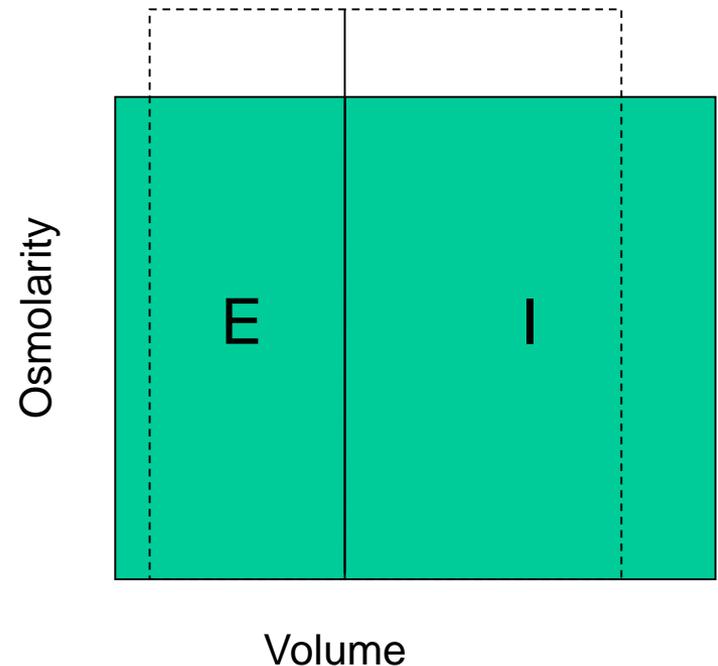
## Water diuresis

Diabetes insipidus

Central

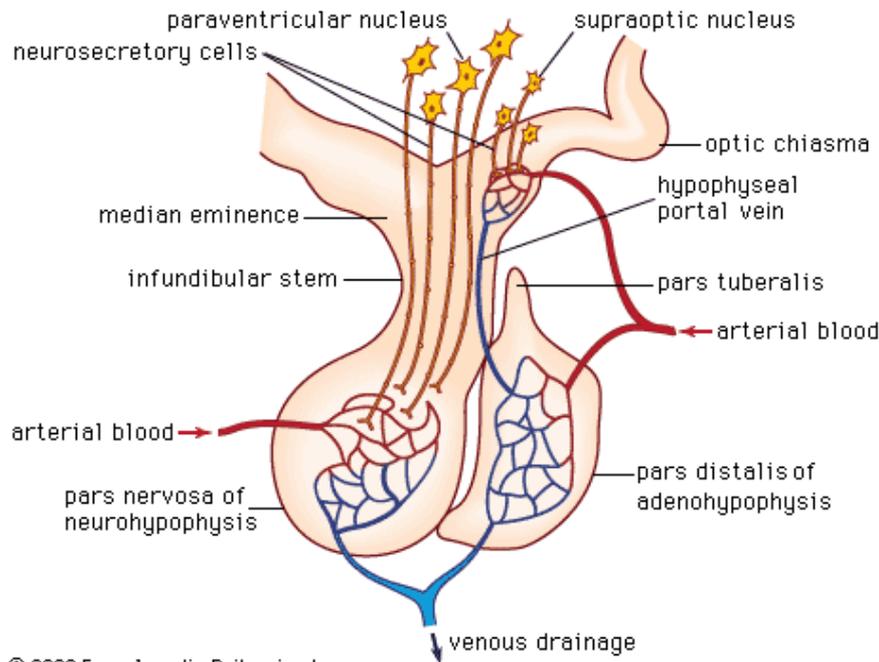
Nephrogenic

Andy had a water diuresis!

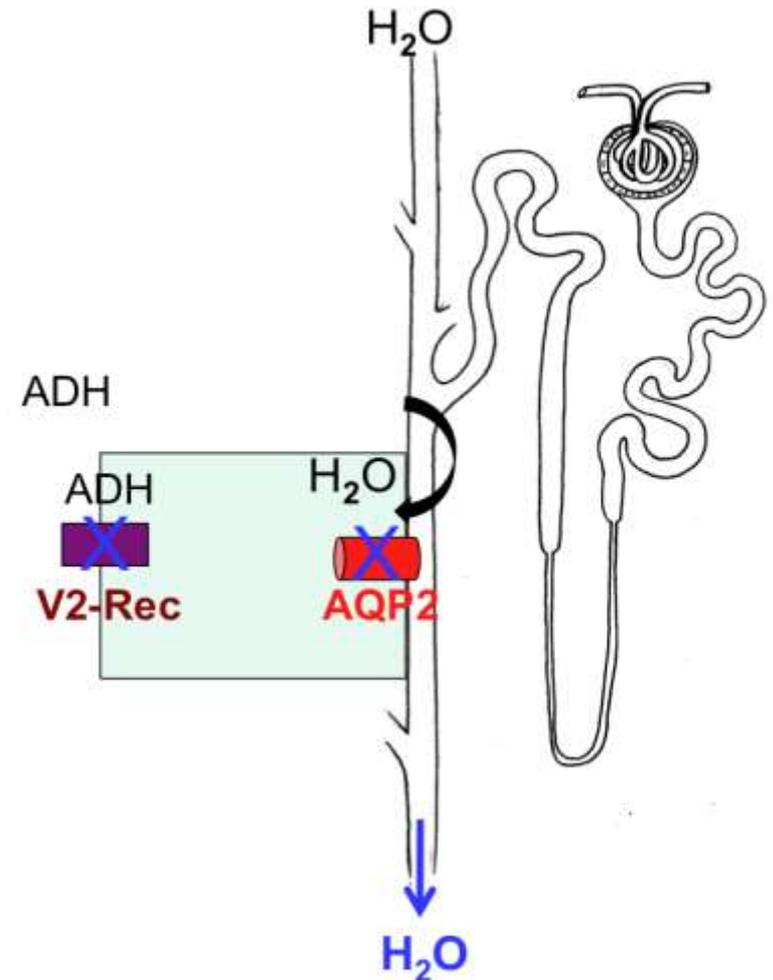


Andy had a sister who “peed” about half as much as he did

# ADH release disturbed (central diabetes insipidus)



ADH is there, kidney does not react (nephrogenic DI)



Infections, tumor,  
trauma, postpartum

# Chief has good experience with this treatment!

64 year-old patient, hypertensive, stroke, has PEG.  
Transferred to convalescence with HCTZ Na 134

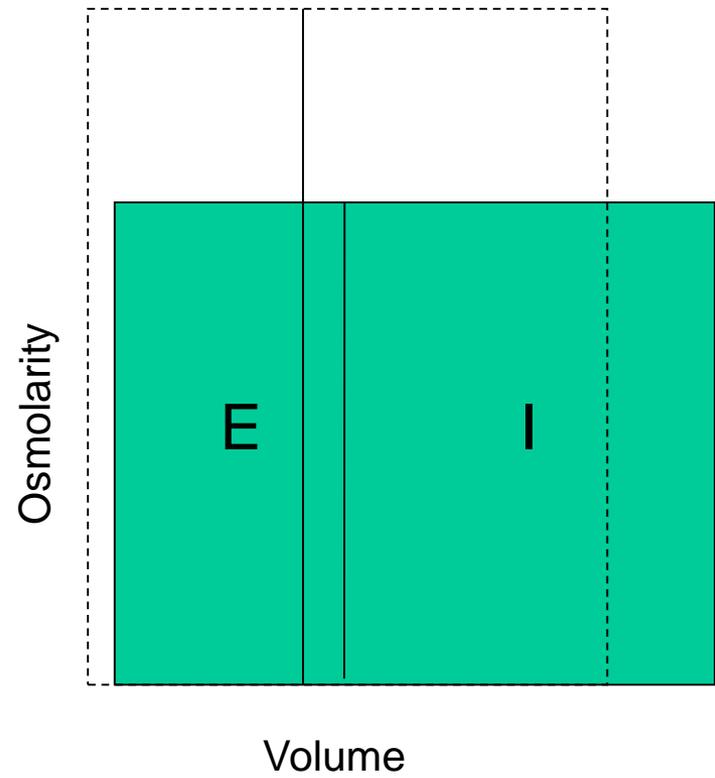
10 days later transferred by to us with Na 199, K 2.1  
Cl 168 (all mmol/L), mild hyperchloremic acidosis,  
creatinine 299  $\mu$ mol/L

BP 140/96 mm Hg. Patient comatose, findings  
were rales and copious edema

How do you get hypernatremia and edema?

Patient had received teaspoons of salt per PEG for  
“hyponatremia”, 330 mmol/day daily for 10 days.

# Salt poisoning by accident or by physicians



Urine Na 133, urine K 46 (mmol/L)

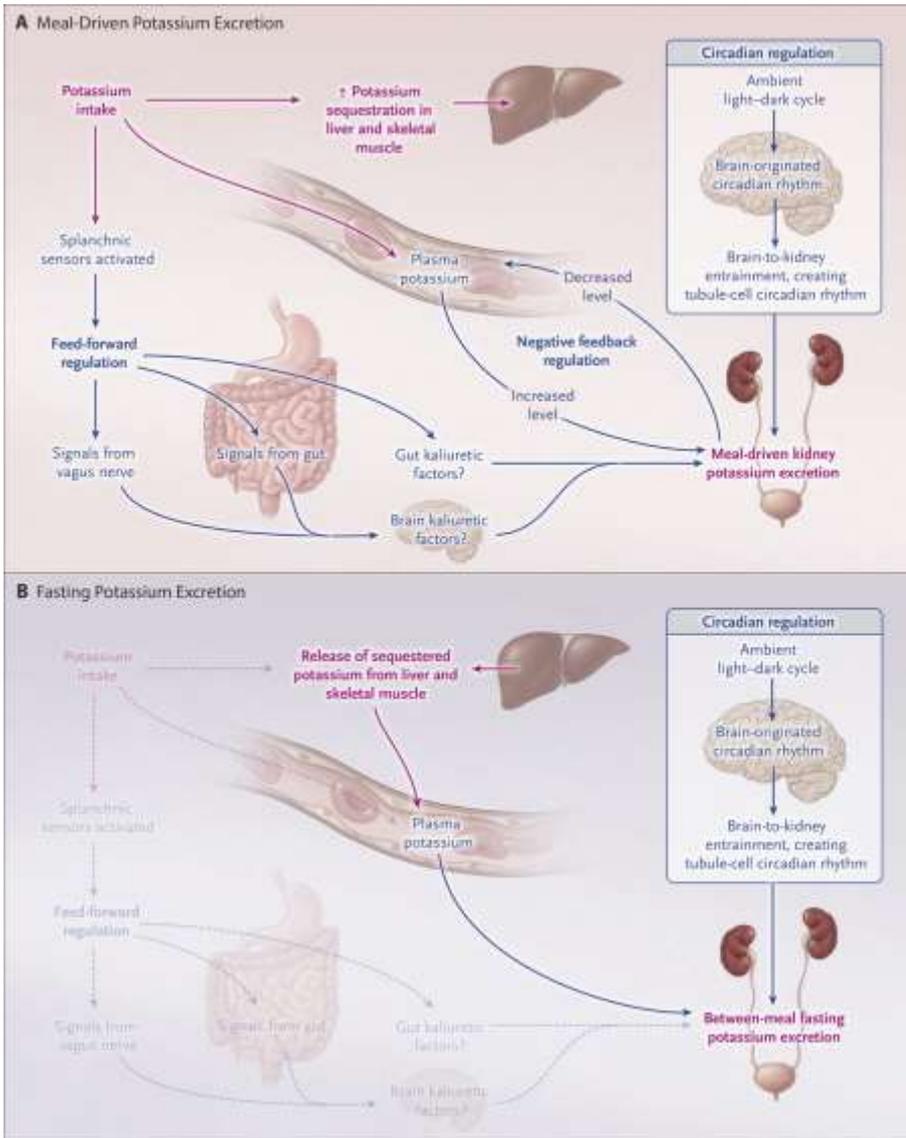
$$\text{Water „deficit“} = 0.6 \times \text{Bwt (kg)} \times \left( \frac{\text{Na measured}}{140} \right) - 1$$

= 14 L (but we could not just give this patient 14 L water)

Since patient was extracellular volume expanded we relied on slow hemofiltration and careful “free-water” (5% glucose) infusions

To our surprise, he survived little worse for wear

**Salt poisoning**



98% K inside cells (ICV)

2% (70 mmol) ECV

5 bananas and you die!

“Feed forwards”  
“Feed backwards”

N Engl J Med 2015

# Paroxysmal Hyperkalemia

50 year-old woman developed “hyperkalemic” episodes 3 years earlier. Her creatinine was 2.5 mg/dL. She had interstitial nephritis but was stable.

She could “sense” the attacks and would appear in emergency with potassium >8 mmol/L. Na and Cl would be unremarkable. 24 hour urine showed normal Na and K excretion (80 mmol/24 h)

Episodes were dramatic with tetraplegia and strange EKGs

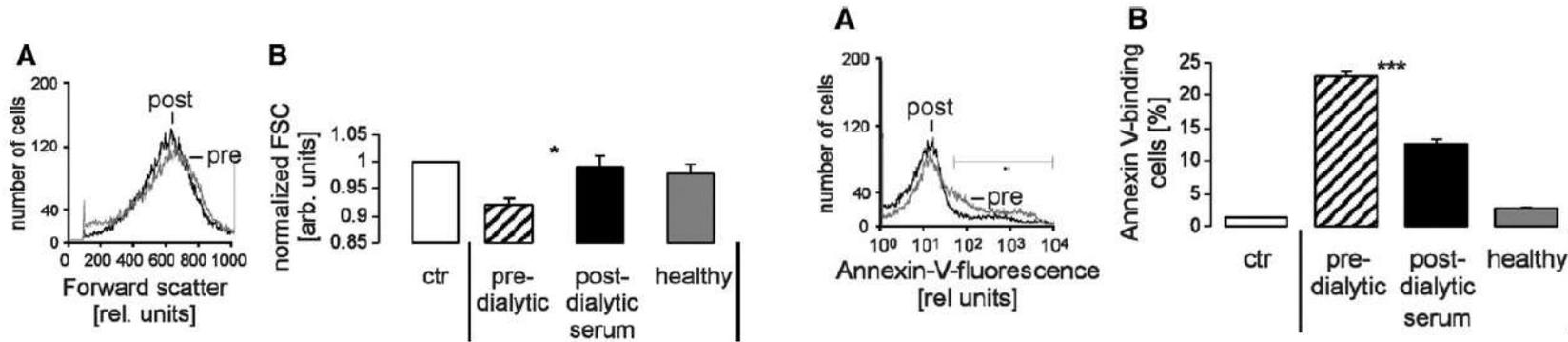
She was treated with furosemide, Resonium®, glucose, insulin, HCO<sub>3</sub>, acute dialysis, and then a defibrillator was placed.

She had gene diagnostics for periodic paralyses of various sorts.

A diagnosis of eryptosis was considered

Finally FCL was consulted.

# She was extensively worked up for eryptosis



Eryptosis is erythrocyte apoptosis. It could release K

I asked about her acid-base status (it is OK!!)

But what about during the attacks?

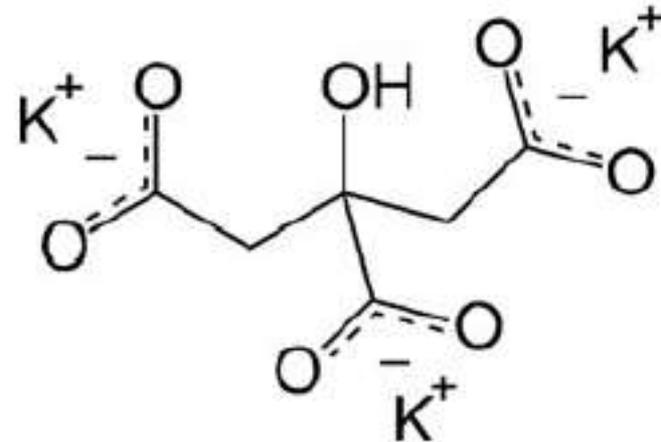
Hmmmm?

Finally, she came in with an attack and vomited:

Vomit K 190 mmol/L, vomit Na 48 mmol/L,

vomit citrate (some time later) >5 g/L

The patient denied “everything” but the episodes stopped.



Die trinkfertige Lösung enthält: 40,0 mmol Kalium-Ionen (entsprechend 1,56 g Kalium) mindestens 13,3 mmol Citrat-Ionen

## Eryptosis plus Munchhausen's syndrome

# Our neurologists deal with tetraparesis

56 year-old patient from the “Wedding”  
cannot get up from the couch

He shows up in our emergency and is seen by a surgeon

“Must be a neurological disease” They keep him for 8 days

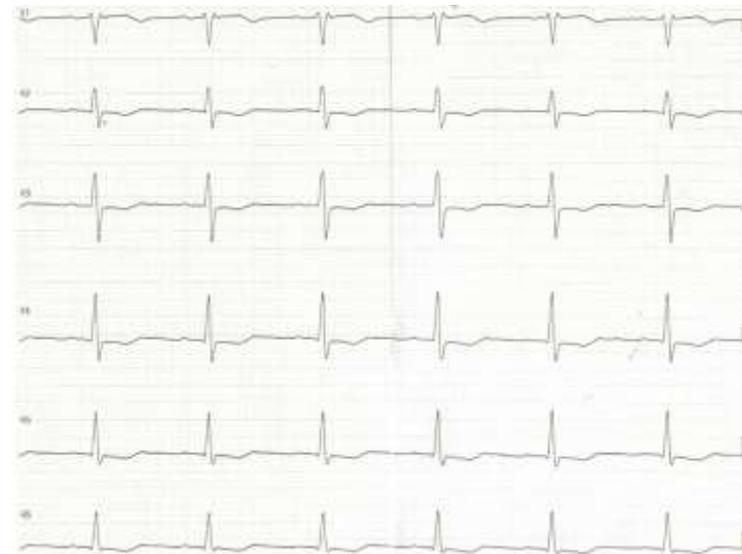
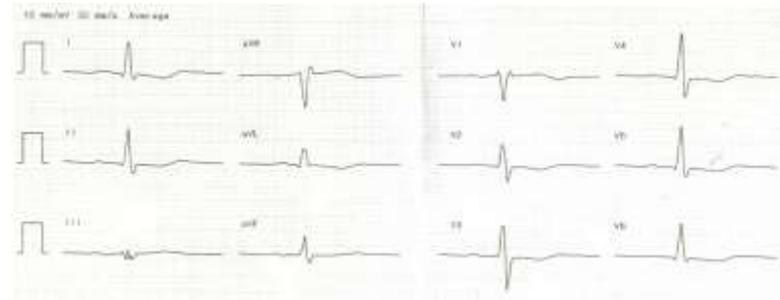
Neurologists find normal BP and tetraparesis

They do lumbar puncture, CT, MRI, EMG, multiple  
measurements are made. They suspect thiamine deficiency

Diagnoses: sensory-motor demyelinating polyneuropathy,  
carpal tunnel syndrome, alcoholism (intake 150 g/day).

Their quality-control rating was “excellent at 100%”

Patient returns to couch but is not better, returns to emergency  
This time he is seen by internal medicine



Na 145, Cl 111, K 1.5 mmol/L

Phosphate 0.44, Mg 0.86 mmol/L

Patient is sent to ICU and they pour in KCl

Next morning: “would you guys be interested in a patient with a K 1.5 mmol/L?” We allowed that we would.

pH	7,396	
pO <sub>2</sub>	92,3	mmHg
pCO <sub>2</sub>	24,0	mmHg
Oxymetrie Ergebnis		
↓ ctHb	11,8	g/dL
Hct <sub>C</sub>	36,4	%
sO <sub>2</sub>	98,2	%
FO <sub>2</sub> Hb	92,2	%
↑ FCOHb	5,0	%
FHHb	1,7	%
↑ FMetHb	1,1	%
p50 <sub>e</sub>	24,36	mmHg
p50(st) <sub>d</sub>	26,84	mmHg
p50(T) <sub>e</sub>	24,36	mmHg
Elektrolyt Ergebnis		
cK <sup>+</sup>	3,6	mmol/L
cNa <sup>+</sup>	137	mmol/L
cCa <sup>2+</sup>	1,27	mmol/L
↑ cCl <sup>-</sup>	115	mmol/L
Metabolit Ergebnis		
cGlu	116	mg/dL
cLac	7	mg/dL
Säure Basen Status		
SBE <sub>C</sub>	-9,5	mmol/L
cHCO <sub>3</sub> <sup>-</sup> (P) <sub>C</sub>	14,4	mmol/L

First set of blood gases

Somewhat overcompensated metabolic acidosis without broad anion gap.

Hypokalemia looks improved

Urine pH 7.0, Na 65, K 15, Cl 67 (mmol/L)

Urine anion gap was negative (ammonium chloride excretion)

Transtubular K gradient 12

At discharge 14 days later:

Na 138, K 3.6, Ca 2.41, Cl 99 mmol/l

Crea 0.76 mg/dl, Urea 17 mg/dl,

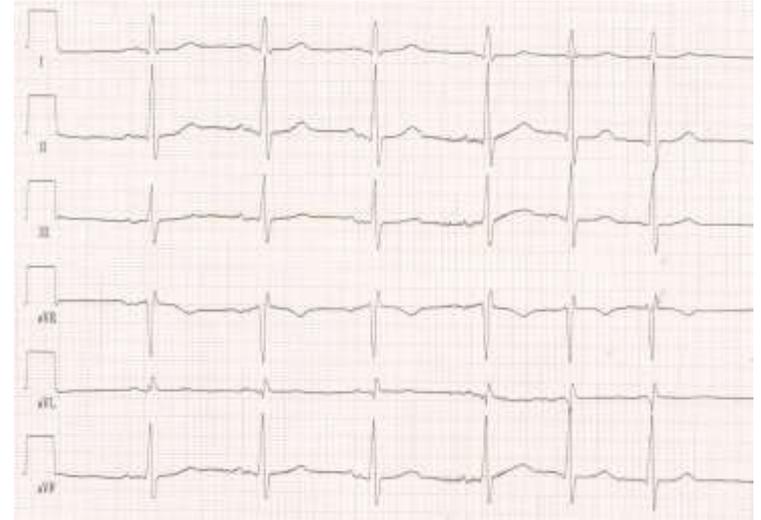
Phos 1.18, Mg 0.78 mmol/l

BGA: pH 7.43 pO<sub>2</sub> 84,7, pCO<sub>2</sub> 31.5, HCO<sub>3</sub> 22.5

Urine: pH 7.0, Na 92, K 24, CL 110 mmol/l (negative UAG)

TTKG 4.4

Tub Phosphate reabsorption 85.5



# Sodium

# Potassium

Hypovolemic hyponatremia

Decreased Total-body potassium

Hypervolemic hyponatremia

$$Na_s = (Na_e + K_e)/TBW$$

“Relatively-preserved” Total-body potassium

Euvolemic hyponatremia

TBW

Decreased

Normal

Increased

Hypovolemic hyponatremia

Euvolemic hyponatremia

Hypervolemic hyponatremia

Water

# I refer you to recent N Engl J Med reviews (2014-15)

- Hyponatremia is always a defect in water excretion. Check to see if a dilute urine can be made (UNa **and UK**).
- ADH presence can most easily be determined by urine osmolarity. It can be calculated.
- Extracellular volume assessment seems simple but is inexact. Nonetheless, edema (yes/no), orthostatic heart-rate and BP changes must always be documented.
- How physicians think is as important as how they do not. We need genuine “instructors” in emergency and on the wards.
- Otherwise, we will be the victims of specialists